## <u>Problem Set 9: Lewis Structures, Electron Pair Geometry, and Molecular Geometry</u>

1.	Draw a Lewis structure to represent each of the following molecules.	
	a.	$NH_3$
	b.	H <sub>2</sub> O
	C.	CCI <sub>4</sub>
	d.	$CO_2$
	e.	HCN
	f.	PCI <sub>3</sub>
	g.	SO <sub>2</sub>
2.	For	r each of the molecules in problem #1, determine the electron pair geometry and the elecular geometry.